

2022.4.13 Lecture 2

Lecture on

10:25 – 11:55

# Magnetic Properties of Materials

磁性 (Magnetism)

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## Chapter1 Basic Notions of Magnetism

Classical pictures of magnetic moments in materials:

- Magnetic charges
- Circular currents

Experimental methods to measure magnetization

Paramagnetic and diamagnetic terms in classical magnetization

Breakdown of classical magnetism: cancellation of paramagnetic and diamagnetic terms (Bohr-van Leeuwen theorem)

Introduction of spin angular momentum by relativistic quantum mechanics

1. Spin-orbit interaction
2. Magnetism in quantum theory

## Chapter 2 Magnetism in localized systems

1. Spherical potential
2. Larmor precession
3. Magnetism of inert gas
4. LS multiplex ground state of open shell ions and Hund's rule

# Magnetic moment of electron spin

Dirac eq. with electromagnetic field  $i\hbar \frac{\partial \psi}{\partial t} = [c\boldsymbol{\alpha}(\mathbf{p} + e\mathbf{A}) + \beta m - e\phi] \psi$

$$\left[ \left( i\hbar \frac{\partial}{\partial t} + e\phi \right) - c \sum_{j=x,y,z} \alpha_j \left( -i\hbar \frac{\partial}{\partial r_j} + eA_j \right) - \beta mc^2 \right] \psi = 0$$

Operation from left:  $i\hbar \frac{\partial}{\partial t} + e\phi + c \sum_{j=x,y,z} \alpha_j \left( -i\hbar \frac{\partial}{\partial r_j} + eA_j \right) + \beta mc^2$

We obtain

$$\left[ \left( i\hbar \frac{\partial}{\partial t} + e\phi \right)^2 - c^2(\mathbf{p} + e\mathbf{A})^2 - m^2c^4 + ic\hbar e(\boldsymbol{\alpha} \cdot \mathbf{E}) + i\hbar c^2 e(\alpha_x \alpha_y B_z + \alpha_y \alpha_z B_x + \alpha_z \alpha_x B_y) \right] \psi = 0$$

Because  $\alpha_x \alpha_y = i\sigma_z^{(4)}$ ,  $\alpha_y \alpha_z = i\sigma_x^{(4)}$ ,  $\alpha_z \alpha_x = i\sigma_y^{(4)}$

$$\left[ \left( i\hbar \frac{\partial}{\partial t} + e\phi \right)^2 - c^2(\mathbf{p} + e\mathbf{A})^2 - m^2c^4 + ic\hbar e(\boldsymbol{\alpha} \cdot \mathbf{E}) - \hbar c^2 e \boldsymbol{\sigma} \cdot \mathbf{B} \right] \psi = 0$$

# Magnetic moment of electron spin (2)

Stationary solution:  $\psi(\mathbf{r}, t) = \exp(-i\epsilon t/\hbar)\varphi(\mathbf{r})$

$$\left[ (\epsilon + e\phi)^2 - c^2(\mathbf{p} + e\mathbf{A})^2 - m^2c^4 + ic\hbar e(\boldsymbol{\alpha} \cdot \mathbf{E}) - \hbar c^2 e \boldsymbol{\sigma} \cdot \mathbf{B} \right] \varphi = 0$$

$$\phi = 0, \mathbf{E} = 0$$

Low energy  
expansion

$$\epsilon = mc^2 + \delta \quad \text{We take first order in } \frac{\delta}{mc^2}$$

$$\left[ \frac{1}{2m}(\mathbf{p} + e\mathbf{A})^2 + \frac{e\hbar}{2m} \boldsymbol{\sigma} \cdot \mathbf{B} \right] \varphi = \delta \varphi$$

Bohr magneton  $\mu_B \equiv \frac{e\hbar}{2m} \approx 9.274 \times 10^{-24} \text{ JT}^{-1}$

$$\frac{e\hbar}{2m} \boldsymbol{\sigma} \cdot \mathbf{B} = \mu_B \boldsymbol{\sigma} \cdot \mathbf{B} = 2\mu_B \mathbf{s} \cdot \mathbf{B}$$

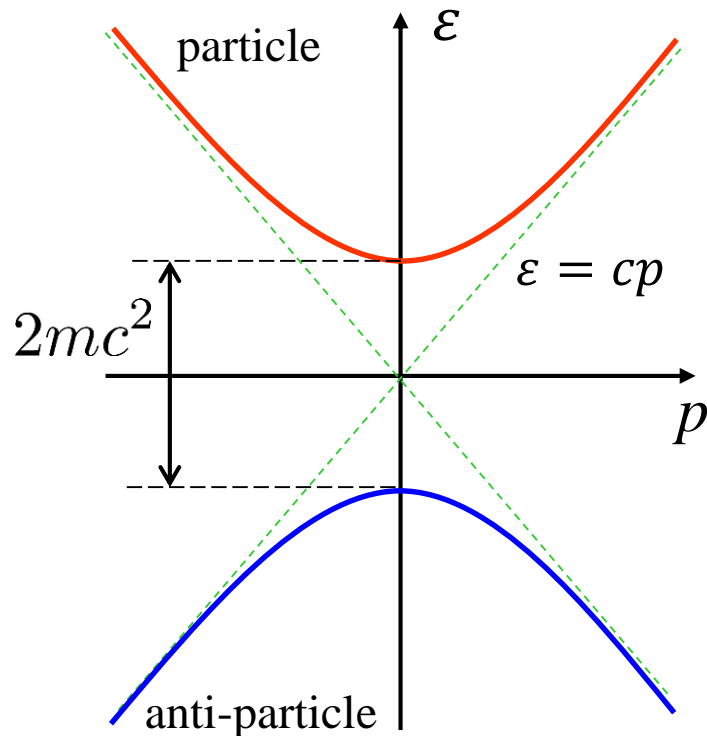
Therefore the magnetic moment is  $-2\mu_B \mathbf{s}$

# Two-component separation approximation

Stationary Dirac equation  $[c\boldsymbol{\alpha}\mathbf{p} + mc^2\beta]\varphi = \epsilon\varphi$

Pauli representation  $\alpha_k = \begin{pmatrix} 0 & \sigma_k \\ \sigma_k & 0 \end{pmatrix}, \quad \beta = \begin{pmatrix} I & 0 \\ 0 & -I \end{pmatrix} \quad 4 \times 4 \text{ matrices}$

When the particle sits still:  $\epsilon = \pm mc^2$



+ corresponds to  $I$ , - corresponds to  $-I$  in  $\beta$

→ upper two laws: particle, lower two: anti-particle (?)

Finite momentum  $p$  requires correction.

$$\tan 2\theta = \frac{p}{mc} \quad \psi_{\uparrow} = e^{i(kz - \omega t)} \begin{pmatrix} \cos \theta \\ 0 \\ \sin \theta \\ 0 \end{pmatrix} \quad \text{Leak to lower laws}$$

# Spin-orbit interaction

Stationary equation  $(c\boldsymbol{\alpha} \cdot \mathbf{p} + \beta mc^2 + V)\varphi = \epsilon\varphi$

Two-component approximation  $\varphi = \begin{pmatrix} \varphi_A \\ \varphi_B \end{pmatrix}$

Simultaneous equations  $\begin{cases} \boldsymbol{\sigma} \cdot \mathbf{p}\varphi_B = c^{-1}(\delta - V)\varphi_A, \\ \boldsymbol{\sigma} \cdot \mathbf{p}\varphi_A = c^{-1}(\delta - V + 2mc^2)\varphi_B. \end{cases} \quad \delta = \epsilon - mc^2$

Erase of  $\varphi_B$   $c^{-2}\boldsymbol{\sigma} \cdot \mathbf{p}(\delta - V + 2mc^2)^{-1}\boldsymbol{\sigma} \cdot \mathbf{p}\varphi_A = (\delta - V)\varphi_A$

Low velocity ( $p \ll mc$ ) expansion  $c^2(\delta - V + 2mc^2)^{-1} \approx \frac{1}{2m} \left[ 1 - \frac{\delta - V}{2mc^2} + \dots \right]$

Normalization condition  $\langle \varphi | \varphi \rangle = \langle \varphi_A | \varphi_A \rangle + \langle \varphi_B | \varphi_B \rangle = 1$

# Spin-orbit interaction (2)

Introduction of magnetic field  $\mathbf{p} \rightarrow \mathbf{p} + e\mathbf{A}$

Correction due to leakage  $\langle \varphi_B | \varphi_B \rangle = \left\langle \varphi_A \left| \left[ \frac{p^2 + e\hbar\boldsymbol{\sigma} \cdot \mathbf{B}}{4m^2c^2} \right] \right| \varphi_A \right\rangle = O\left(\frac{v^2}{c^2}\right)$

Corrected two-component wavefunction  $\varphi_a = \left( 1 + \frac{p^2 + e\hbar\boldsymbol{\sigma} \cdot \mathbf{B}}{8m^2c^2} \right) \varphi_A$

Pauli two-component approximation

$$\left[ \frac{p^2}{2m} + V + \frac{e\hbar}{2m} \boldsymbol{\sigma} \cdot \mathbf{B} - \frac{e\hbar \boldsymbol{\sigma} \cdot \mathbf{p} \times \mathbf{E}}{4m^2c^2} - \frac{e\hbar^2}{8m^2c^2} \nabla \cdot \mathbf{E} - \frac{p^4}{8m^3c^2} - \frac{e\hbar p^2}{4m^3c^2} \boldsymbol{\sigma} \cdot \mathbf{B} - \frac{(e\hbar B)^2}{8m^3c^2} \right] \varphi_a = \delta\varphi_a$$

Zeeman      Spin-orbit interaction



# Quantum Mechanical Treatment of Magnetism

$$\mathcal{H} = \sum_n \left[ \frac{1}{2m} (\mathbf{p}_n + e\mathbf{A}(\mathbf{r}_n))^2 + U(\mathbf{r}_n) + g\mu_B \mathbf{s}_n \cdot \mathbf{B} \right] + V(\mathbf{r}_1, \mathbf{r}_2, \dots)$$

↓
↓  
 Nucleus potential      **g-factor**

Symmetric gauge     $\mathbf{A}(\mathbf{r}_n) = (\mathbf{B} \times \mathbf{r}_n)/2$

$$\mathcal{H} = \sum_n \left[ \frac{\mathbf{p}_n^2}{2m} + U(\mathbf{r}_n) \right] + V(\mathbf{r}_1, \mathbf{r}_2, \dots) \quad \dots \quad \mathcal{H}_0$$

$$\hbar \mathbf{l}_n \equiv \mathbf{r}_n \times \mathbf{p}_n \quad + \mu_B \sum_n (\mathbf{l}_n + g\mathbf{s}_n) \cdot \mathbf{B} \quad \dots \quad \mathcal{H}_1$$

$$+ \frac{e^2}{8m} \sum_n \{ r_n^2 B^2 - (\mathbf{B} \cdot \mathbf{r}_n)^2 \} \quad \dots \quad \mathcal{H}_2$$

# Magnetic moment

Commutation relations

$$\left\{ \begin{array}{l} [r_{n\alpha}, p_{n\beta}] = r_{n\alpha}p_{n\beta} - p_{n\beta}r_{n\alpha} = i\hbar\delta_{\alpha\beta} \quad (\alpha, \beta = x, y, z) \\ [s_{n\alpha}, s_{n\beta}] = i s_{n\gamma} \quad (\alpha, \beta, \gamma = x, y, z \text{ (cyclic)}) \\ [l_{n\alpha}, l_{n\beta}] = i l_{n\gamma} \quad (\alpha, \beta, \gamma = x, y, z \text{ (cyclic)}) \end{array} \right.$$

Magnetic moment

$$\begin{aligned} \mu &= -\frac{\partial \mathcal{H}}{\partial \mathbf{B}} = -\mu_B \sum_n (\mathbf{l}_n + g\mathbf{s}_n) - \frac{e^2}{4m} \sum_n \{r_n^2 \mathbf{B} - \mathbf{r}_n (\mathbf{r}_n \cdot \mathbf{B})\} \\ &= -\mu_B \sum_n (\mathbf{l}_n + g\mathbf{s}_n) - \frac{e^2}{4m} \sum_n (\mathbf{r}_n \times (\mathbf{B} \times \mathbf{r}_n)) \end{aligned}$$

Paramagnetic Diamagnetic

This expression does not have drastic changes other than spin magnetic moment.  
However ...

# Comment: Spins of nucleons

Protons, Neutrons, Muons have spins.



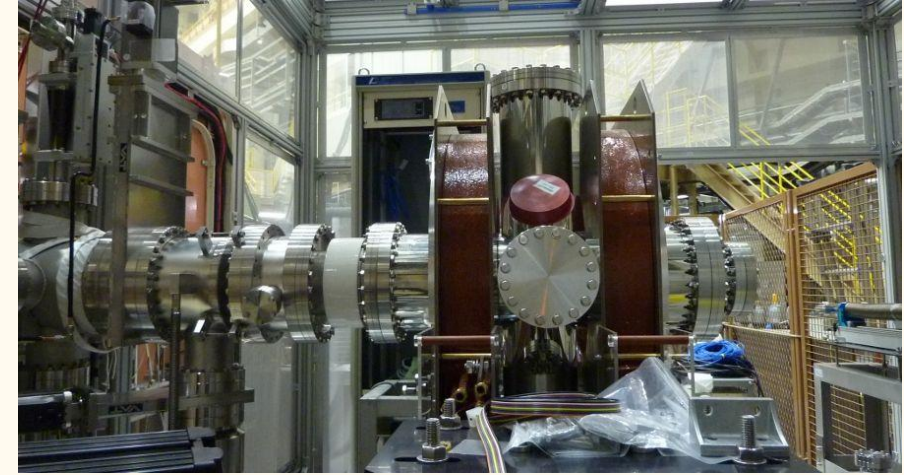
MRI

NMR



J-PARC

Neutron diffraction



KEK

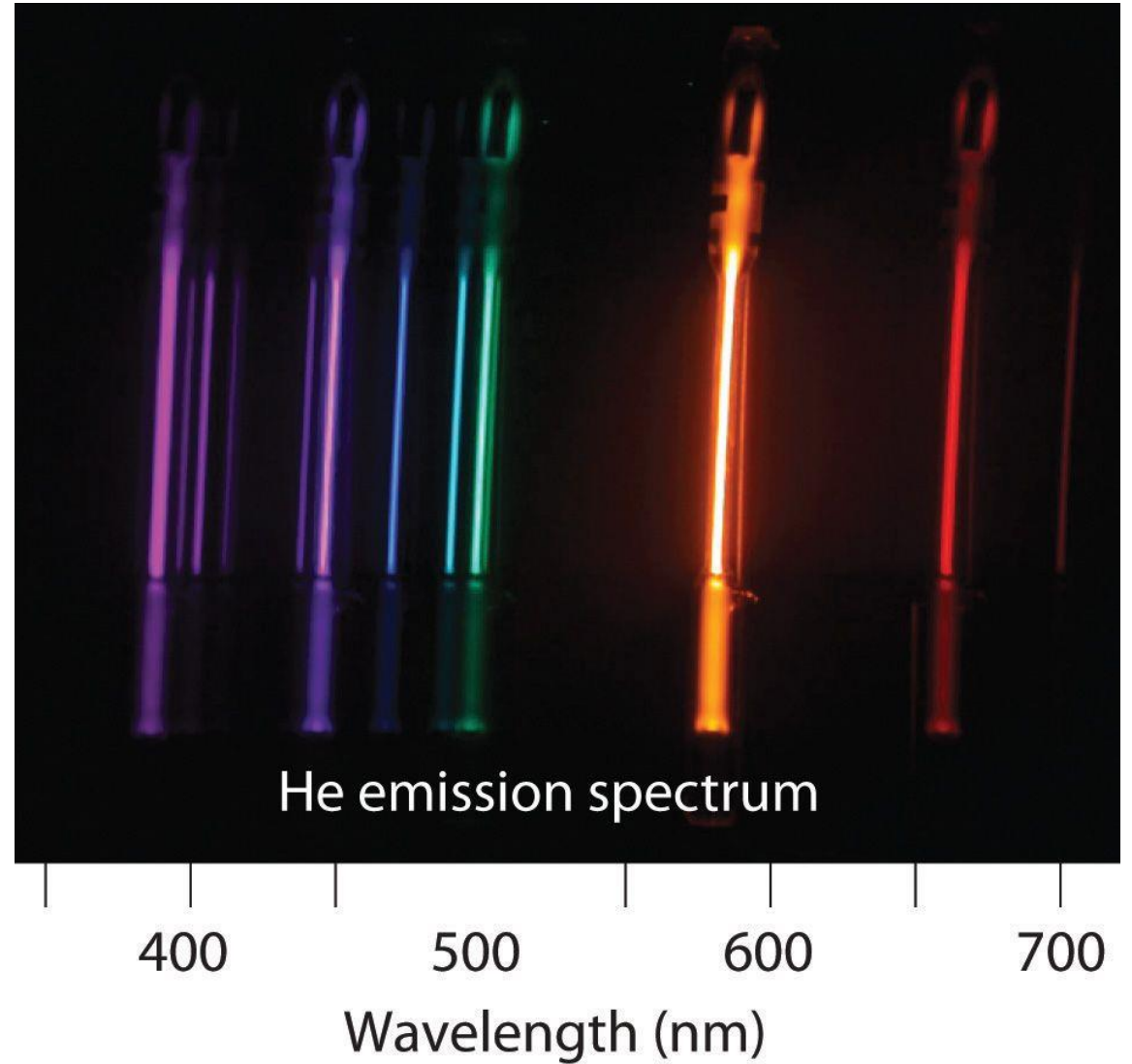
$\mu$ SR

# Chapter 2

## Magnetism of Localized Electrons



Star birth



# Second quantization

$|\mathbf{n}\rangle = |n_1, n_2, \dots\rangle$  Number representation  
(index the state with number of particles occupying basis states)

$|0\rangle$  Vacuum

$a_j^\dagger |0\rangle = |1_j\rangle$  Creation operator of  $j$ -th state  
(Hermitian conjugate: annihilation operator)

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Fermion:

anti-commutation relation

$$[a_i, a_j]_+ = [a_i^\dagger, a_j^\dagger]_+ = 0, \quad [a_i, a_j^\dagger]_+ = \delta_{ij}$$

number operator

$$\hat{n}_j \equiv a_j^\dagger a_j \quad \hat{n}_j |\mathbf{n}\rangle = n_j |\mathbf{n}\rangle$$

Boson: commutation relation

$$[b_i, b_j] = [b_i^\dagger, b_j^\dagger] = 0, \quad [b_i, b_j^\dagger] = \delta_{ij}$$

$$|n_j\rangle = \frac{1}{\sqrt{n_j!}} (a_j^\dagger)^{n_j} |0\rangle$$

# Operators in second quantization representation

Multiparticle operator  $\mathcal{F}(\mathbf{r}_1, \mathbf{r}_2, \dots) = \sum_i f(\mathbf{r}_i)$

Slater determinant  $|\psi_{1,2,\dots}\rangle$

$$\langle \psi_{m_1, m_2, \dots} | \mathcal{F} | \psi_{n_1, n_2, \dots} \rangle = \sum_i \langle \psi_{m_1, m_2, \dots} | f(\mathbf{r}_i) | \psi_{n_1, n_2, \dots} \rangle$$

Second quantization

$$F = \sum_{mn} \langle m | f | n \rangle a_m^\dagger a_n$$

Particle statistics

$$\langle m | f | n \rangle = \int d\mathbf{r} \phi_m^*(\mathbf{r}) f(\mathbf{r}) \psi_n(\mathbf{r})$$

Annihilation and creation operator (anti-)commutation relations

$$\langle \psi_{m_1, m_2, \dots} | \mathcal{F} | \psi_{n_1, n_2, \dots} \rangle = \langle \mathbf{m} | F | \mathbf{n} \rangle$$

$$G = \frac{1}{2} \sum_{klmn} \langle kl | g | mn \rangle a_k^\dagger a_l^\dagger a_n a_m$$

# Electrons in a central force potential

$$\mathcal{H}_L = \mathcal{H}_{L0} + \mathcal{H}_C + \mathcal{H}_{\text{SOI}} + \mathcal{H}_{\text{CF}}$$

Localized system

crystal field

spin-orbit interaction

mutual Coulomb interaction

single-electron (non-interaction)

$$\mathcal{H}_{L0} = \sum_j \left[ \frac{\mathbf{p}_j^2}{2m} + V_{\text{sp}}(r_j) \right] \quad \text{Electrons in a central force (spherical) potential}$$

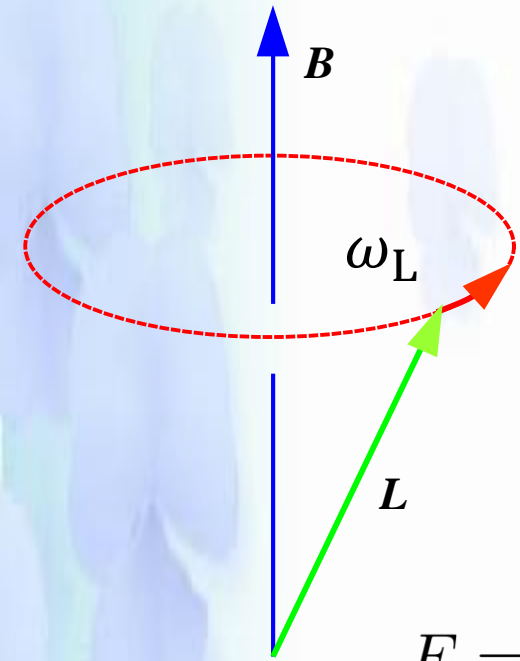
Eigenfunction in polar coordinate:  $(r, \theta, \varphi)$        $\psi_{nlm}(\mathbf{r}) = R_{nl}(r)Y_{lm}(\theta, \varphi)$

Radial wavefunction  $R_{nl}(r) = b_{nl}\rho^l e^{-\rho/2} L_{n+l-1}^{2l+1}(\rho), \quad \rho \equiv \frac{2}{n} \frac{r}{a_0}$

Eigen energy  $\epsilon_{nl} = -\frac{R_\infty}{n^2}, \quad R_\infty = \frac{me^4}{8\epsilon_0 h^3 c}$

$$\mathcal{H}_{L0} = \sum_{nl} \epsilon_{nl} \sum_{m\sigma} a_{nlm\sigma}^\dagger a_{nlm\sigma}$$

# Larmor precession



Coulomb potential  $V_{\text{sp}}(r_j) = -\frac{Ze^2}{4\pi\epsilon_0} \frac{1}{r_j}$

Total orbital angular momentum  $\hbar\mathbf{L} = \hbar \sum_i \mathbf{l}_i$

$$\mathcal{H}_1 = \mu_B \mathbf{L} \cdot \mathbf{B} = \mu_B L_z B$$

Directional quantization  $L_z = M : -L, -L + 1, \dots, L - 1, L$

$$E = E_0 + \mu_B M B \equiv E_0 + \hbar\omega_L M, \quad \omega_L \equiv \frac{\mu_B B}{\hbar} = \frac{eB}{2m} \text{ (Larmor frequency)}$$

Heisenberg equation  $\frac{d\mathbf{L}}{dt} = \frac{1}{i\hbar} [\mathbf{L}, \mathcal{H}_0 + \mathcal{H}_1 + \mathcal{H}_2]$

**Larmor precession**  $L_x(t) = L_0 \cos(\omega_L t + \theta_0), \quad L_y(t) = L_0 \sin(\omega_L t + \theta_0)$

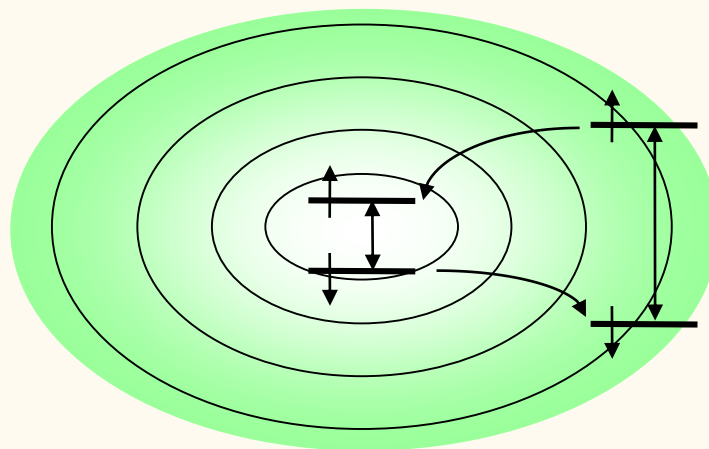
In the case of spin: g-factor  $\omega_L = g \frac{eB}{2m} \approx \frac{eB}{m}$



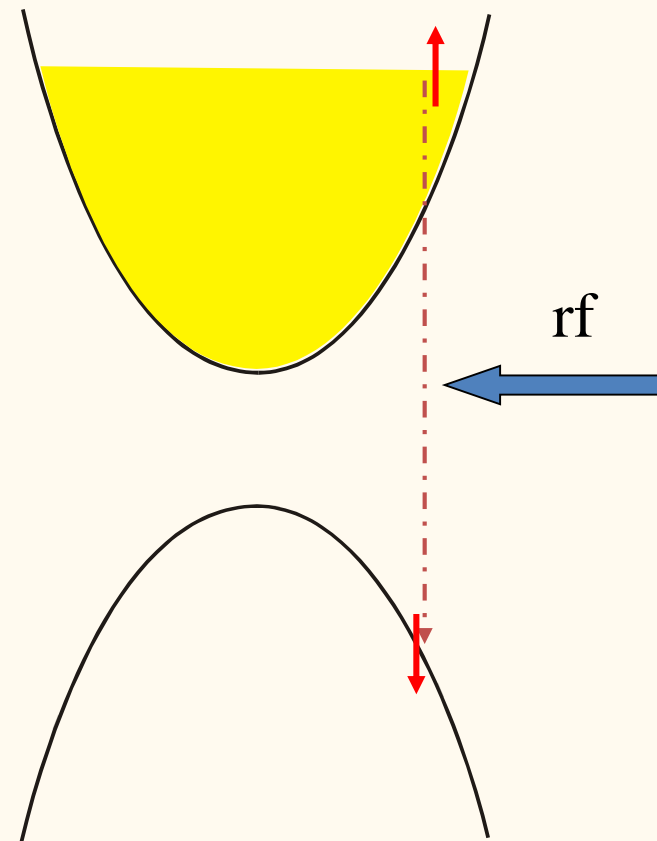
# Magnetism of inert gases



Star birth



Magnetic trap



Evaporation cooling

Total angular momentum

$$\mathbf{L} = \sum_j \mathbf{l}_j = \sum_{\sigma} \sum_{mm'} \langle m | \mathbf{l} | m' \rangle_{nl} a_{m\sigma}^{\dagger} a_{m'\sigma},$$

$$\mathbf{S} = \sum_j \mathbf{s}_j = \sum_m \sum_{\sigma\sigma'} \left( \frac{\sigma}{2} \right)_{\sigma\sigma'} a_{m\sigma}^{\dagger} a_{m\sigma'},$$

$$\mathbf{J} = \sum_j \mathbf{j}_j = \mathbf{L} + \mathbf{S}$$

# Magnetism of inert gases

Inert gases: Closed shell structure  $L = S = 0$  due to quantization!

Residual is the dielectric term: 
$$\begin{aligned} \mu_{\text{dia}} &= -\frac{e^2}{4m} \sum_n [\mathbf{r}_n \times (\mathbf{B} \times \mathbf{r}_n)] \\ &= -\frac{e}{2} \sum_n [\mathbf{r}_n \times (\boldsymbol{\omega}_L \times \mathbf{r}_n)] = -\frac{\mu_B}{\hbar} \sum_n [\mathbf{r}_n \times (m\mathbf{v}_n)] \end{aligned}$$

Larmor rotation angular momentum

$$\mu_d = -\frac{e^2}{4m} \langle x^2 + y^2 \rangle B = -\frac{e^2}{6m} \langle r^2 \rangle B$$

$$\chi = -\frac{N_A Z e^2 \langle r^2 \rangle}{6m} \quad \text{Moll susceptibility}$$

$$\frac{\langle r^2 \rangle}{a_B^2} \sim ??$$

| Z  | Element | Susceptibility         |
|----|---------|------------------------|
| 2  | He      | $-1.9 \times 10^{-6}$  |
| 10 | Ne      | $-7.2 \times 10^{-6}$  |
| 18 | Ar      | $-19.4 \times 10^{-6}$ |
| 36 | Kr      | $-28 \times 10^{-6}$   |
| 54 | Xe      | $-43 \times 10^{-6}$   |

# PERIODIC TABLE OF ELEMENTS



|   |  |   |   |   |  |   |  |  |  |   |   |   |  |  |  |  |  |                                       |
|---|--|---|---|---|--|---|--|--|--|---|---|---|--|--|--|--|--|---------------------------------------|
| 1<br><b>H</b><br>Hydrogen<br>Nonmetal       |  |   |   |   |  |   |  |  |  |   |   |   |  |  |  |  | 2<br><b>He</b><br>Helium<br>Noble Gas      |                                       |
| 3<br><b>Li</b><br>Lithium<br>Alkali Metal   | 4<br><b>Be</b><br>Beryllium<br>Alkaline Earth Metal  |   |   |   |  |   |  |  |  |   |   |   |  |  |  |  |  | 10<br><b>Ne</b><br>Neon<br>Noble Gas  |
| 11<br><b>Na</b><br>Sodium<br>Alkali Metal   | 12<br><b>Mg</b><br>Magnesium<br>Alkaline Earth Metal |   |   |   |  |   |  |  |  |   |   |   |  |  |  |  |  | 18<br><b>Ar</b><br>Argon<br>Noble Gas |
| 19<br><b>K</b><br>Potassium<br>Alkali Metal | 20<br><b>Ca</b><br>Calcium<br>Alkaline Earth Metal   | 21<br><b>Sc</b><br>Scandium<br>Transition Metal | 22<br><b>Ti</b><br>Titanium<br>Transition Metal       | 23<br><b>V</b><br>Vanadium<br>Transition Metal  | 24<br><b>Cr</b><br>Chromium<br>Transition Metal    | 25<br><b>Mn</b><br>Manganese<br>Transition Metal  | 26<br><b>Fe</b><br>Iron<br>Transition Metal      | 27<br><b>Co</b><br>Cobalt<br>Transition Metal      | 28<br><b>Ni</b><br>Nickel<br>Transition Metal        | 29<br><b>Cu</b><br>Copper<br>Transition Metal       | 30<br><b>Zn</b><br>Zinc<br>Transition Metal         | 31<br><b>Ga</b><br>Gallium<br>Post-Transition Metal   | 32<br><b>Ge</b><br>Germanium<br>Metalloid              | 33<br><b>As</b><br>Arsenic<br>Metalloid                | 34<br><b>Se</b><br>Selenium<br>Nonmetal                  | 35<br><b>Br</b><br>Bromine<br>Halogen      | 36<br><b>Kr</b><br>Krypton<br>Noble Gas    |                                       |
| 37<br><b>Rb</b><br>Rubidium<br>Alkali Metal | 38<br><b>Sr</b><br>Strontium<br>Alkaline Earth Metal | 39<br><b>Y</b><br>Yttrium<br>Transition Metal   | 40<br><b>Zr</b><br>Zirconium<br>Transition Metal      | 41<br><b>Nb</b><br>Niobium<br>Transition Metal  | 42<br><b>Mo</b><br>Molybdenum<br>Transition Metal  | 43<br><b>Tc</b><br>Technetium<br>Transition Metal | 44<br><b>Ru</b><br>Ruthenium<br>Transition Metal | 45<br><b>Rh</b><br>Rhodium<br>Transition Metal     | 46<br><b>Pd</b><br>Palladium<br>Transition Metal     | 47<br><b>Ag</b><br>Silver<br>Transition Metal       | 48<br><b>Cd</b><br>Cadmium<br>Transition Metal      | 49<br><b>In</b><br>Indium<br>Post-Transition Metal    | 50<br><b>Sn</b><br>Tin<br>Post-Transition Metal        | 51<br><b>Sb</b><br>Antimony<br>Metalloid               | 52<br><b>Te</b><br>Tellurium<br>Metalloid                | 53<br><b>I</b><br>Iodine<br>Halogen        | 54<br><b>Xe</b><br>Xenon<br>Noble Gas      |                                       |
| 55<br><b>Cs</b><br>Cesium<br>Alkali Metal   | 56<br><b>Ba</b><br>Barium<br>Alkaline Earth Metal    | •   | 72<br><b>Hf</b><br>Hafnium<br>Transition Metal        | 73<br><b>Ta</b><br>Tantalum<br>Transition Metal | 74<br><b>W</b><br>Tungsten<br>Transition Metal     | 75<br><b>Re</b><br>Rhenium<br>Transition Metal    | 76<br><b>Os</b><br>Osmium<br>Transition Metal    | 77<br><b>Ir</b><br>Iridium<br>Transition Metal     | 78<br><b>Pt</b><br>Platinum<br>Transition Metal      | 79<br><b>Au</b><br>Gold<br>Transition Metal         | 80<br><b>Hg</b><br>Mercury<br>Transition Metal      | 81<br><b>Tl</b><br>Thallium<br>Post-Transition Metal  | 82<br><b>Pb</b><br>Lead<br>Post-Transition Metal       | 83<br><b>Bi</b><br>Bismuth<br>Post-Transition Metal    | 84<br><b>Po</b><br>Polonium<br>Metalloid                 | 85<br><b>At</b><br>Astatine<br>Halogen     | 86<br><b>Rn</b><br>Radon<br>Noble Gas      |                                       |
| 87<br><b>Fr</b><br>Francium<br>Alkali Metal | 88<br><b>Ra</b><br>Radium<br>Alkaline Earth Metal    | **  | 104<br><b>Rf</b><br>Rutherfordium<br>Transition Metal | 105<br><b>Db</b><br>Dubnium<br>Transition Metal | 106<br><b>Sg</b><br>Seaborgium<br>Transition Metal | 107<br><b>Bh</b><br>Bohrium<br>Transition Metal   | 108<br><b>Hs</b><br>Hassium<br>Transition Metal  | 109<br><b>Mt</b><br>Meitnerium<br>Transition Metal | 110<br><b>Ds</b><br>Darmstadtium<br>Transition Metal | 111<br><b>Rg</b><br>Roentgenium<br>Transition Metal | 112<br><b>Cn</b><br>Copernicium<br>Transition Metal | 113<br><b>Nh</b><br>Nihonium<br>Post-Transition Metal | 114<br><b>Fl</b><br>Flerovium<br>Post-Transition Metal | 115<br><b>Mc</b><br>Moscovium<br>Post-Transition Metal | 116<br><b>Lv</b><br>Livermorium<br>Post-Transition Metal | 117<br><b>Ts</b><br>Tennessine<br>Halogen  | 118<br><b>Og</b><br>Oganesson<br>Noble Gas |                                       |
|   |  | •   | 57<br><b>La</b><br>Lanthanum<br>Lanthanide            | 58<br><b>Ce</b><br>Cerium<br>Lanthanide         | 59<br><b>Pr</b><br>Praseodymium<br>Lanthanide      | 60<br><b>Nd</b><br>Neodymium<br>Lanthanide        | 61<br><b>Pm</b><br>Promethium<br>Lanthanide      | 62<br><b>Sm</b><br>Samarium<br>Lanthanide          | 63<br><b>Eu</b><br>Europium<br>Lanthanide            | 64<br><b>Gd</b><br>Gadolinium<br>Lanthanide         | 65<br><b>Tb</b><br>Terbium<br>Lanthanide            | 66<br><b>Dy</b><br>Dysprosium<br>Lanthanide           | 67<br><b>Ho</b><br>Holmium<br>Lanthanide               | 68<br><b>Er</b><br>Erbium<br>Lanthanide                | 69<br><b>Tm</b><br>Thulium<br>Lanthanide                 | 70<br><b>Yb</b><br>Ytterbium<br>Lanthanide | 71<br><b>Lu</b><br>Lutetium<br>Lanthanide  |                                       |
|   |  | **  | 89<br><b>Ac</b><br>Actinium<br>Actinide               | 90<br><b>Th</b><br>Thorium<br>Actinide          | 91<br><b>Pa</b><br>Protactinium<br>Actinide        | 92<br><b>U</b><br>Uranium<br>Actinide             | 93<br><b>Np</b><br>Neptunium<br>Actinide         | 94<br><b>Pu</b><br>Plutonium<br>Actinide           | 95<br><b>Am</b><br>Americium<br>Actinide             | 96<br><b>Cm</b><br>Curium<br>Actinide               | 97<br><b>Bk</b><br>Berkelium<br>Actinide            | 98<br><b>Cf</b><br>Californium<br>Actinide            | 99<br><b>Es</b><br>Einsteinium<br>Actinide             | 100<br><b>Fm</b><br>Fermium<br>Actinide                | 101<br><b>Md</b><br>Mendelevium<br>Actinide              | 102<br><b>No</b><br>Nobelium<br>Actinide   | 103<br><b>Lr</b><br>Lawrencium<br>Actinide |                                       |

|          |                      |
|----------|----------------------|
| 1        | Atomic Number        |
| <b>H</b> | Symbol               |
| Hydrogen | Name                 |
| Nonmetal | Chemical Group Block |

# Electronic states in magnetic ions

## Open shell electronic states

Angular momentum  $l$  orbit  $m = -l, -l + 1, \dots, l$

State of many electrons: indexed with  $L$  and  $S$  : state  $(L, S)$  degenerated in the absence of coulomb term

$(L, S)$  term degenerated  $(2L+1)(2S+1)$  : **LS multiplex**

Which state is the ground state?

$$\mathcal{H}_C = \frac{1}{2} \sum_{m_1, \dots, m_4} \sum_{\sigma_1 \sigma_2} \left\langle m_1 m_2 \left| \frac{e^2}{4\pi\epsilon_0 r} \right| m_3 m_4 \right\rangle a_{m_1 \sigma_1}^\dagger a_{m_2 \sigma_2}^\dagger a_{m_3 \sigma_3} a_{m_4 \sigma_4}$$

$$\left\langle m_1 m_2 \left| \frac{e^2}{4\pi\epsilon_0 r} \right| m_3 m_4 \right\rangle = \int d\mathbf{r}_1 d\mathbf{r}_2 u_{m_1}^*(\mathbf{r}_1) u_{m_2}^*(\mathbf{r}_2) \frac{e^2}{4\pi\epsilon_0 |\mathbf{r}_1 - \mathbf{r}_2|} u_{m_3}(\mathbf{r}_2) u_{m_4}(\mathbf{r}_1)$$

# Dominating terms

$$m_1 = m_2 = m_3 = m_4$$

$$\left\langle m_1 m_1 \left| \frac{e^2}{4\pi\epsilon_0 r} \right| m_1 m_1 \right\rangle a_{m_1\uparrow}^\dagger a_{m_1\downarrow}^\dagger a_{m_1\uparrow} a_{m_1\downarrow} = U_0 \sum_m \hat{n}_{m\uparrow} \hat{n}_{m\downarrow} \quad (\hat{n}_{m\sigma} = a_{m\sigma}^\dagger a_{m\sigma})$$

Coulomb repulsion in the same orbit

$$m_1 = m_4 \neq m_2 = m_3$$

$$\frac{1}{2} \sum_{m_1 \neq m_2} U(m_1, m_2) \hat{n}_{m_1} \hat{n}_{m_2} \quad \left( \hat{n}_m = \sum_{\sigma} n_{m\sigma} \right)$$

Coulomb repulsion between different orbits

$$m_1 = m_3 \neq m_2 = m_4$$

Exchange term

$$\frac{1}{2} \sum_{m_1 \neq m_2} \sum_{\sigma_1 \sigma_2} J(m_1, m_2) a_{m_1\sigma_1}^\dagger a_{m_2\sigma_2}^\dagger a_{m_1\sigma_2} a_{m_2\sigma_1}$$

$$= -\frac{1}{2} \sum_{m_1 \neq m_2} J(m_1, m_2) \left( \frac{1}{2} \hat{n}_{m_1} \hat{n}_{m_2} + 2\mathbf{s}_{m_1} \cdot \mathbf{s}_{m_2} \right)$$

# Exchange integral

Spin operator  $\mathbf{s}_m = \sum_{\sigma_1 \sigma_2} \left( \frac{\boldsymbol{\sigma}}{2} \right)_{\sigma_1 \sigma_2} a_{m\sigma_1}^\dagger a_{m\sigma_2}$

Exchange integral  $J(m_1, m_2)$

$$\begin{aligned} J(m_1, m_2) &= \int d\mathbf{r}_1 d\mathbf{r}_2 u_{m_1}^*(\mathbf{r}_1) u_{m_2} \frac{e^2}{4\pi\epsilon_0 |\mathbf{r}_1 - \mathbf{r}_2|} u_{m_1}(\mathbf{r}_2) u_{m_2}^*(\mathbf{r}_2) \\ &= \int d\mathbf{r}_1 d\mathbf{r}_2 u_{m_1}^*(\mathbf{r}_1) u_{m_2} \left[ \int d\mathbf{q} \frac{e^2}{\epsilon_0 q^2} e^{i\mathbf{q} \cdot (\mathbf{r}_1 - \mathbf{r}_2)} \right] u_{m_1}(\mathbf{r}_2) u_{m_2}^*(\mathbf{r}_2) \\ &= \int d\mathbf{q} \frac{e^2}{\epsilon_0 q^2} \left| \int d\mathbf{r}_1 u_{m_1}^*(\mathbf{r}_1) u_{m_2}(\mathbf{r}_1) e^{i\mathbf{q} \cdot \mathbf{r}_1} \right|^2 > 0 \end{aligned}$$

# Hund's rule

## Hund's rule

The ground LS multiplex is determined by the following

1. It should have maximum  $S$ .
2. Under the condition 1., it should have maximum  $L$ .

## 3d transition metal ions

| Element | Configuration  | Ion                                 | Configuration | $L$ | $S$ |
|---------|----------------|-------------------------------------|---------------|-----|-----|
| Sc      | $3d^1 4s^2$    |                                     |               |     |     |
| Ti      | $3d^2 4s^2$    | Ti <sup>3+</sup> , V <sup>4+</sup>  | $3d^1$        | 2   | 1/2 |
| V       | $3d^3 4s^2$    | V <sup>3+</sup>                     | $3d^2$        | 3   | 1   |
| Cr      | $3d^5 4s^1$    | Cr <sup>3+</sup> , V <sup>2+</sup>  | $3d^3$        | 3   | 3/2 |
| Mn      | $3d^5 4s^2$    | Mn <sup>3+</sup> , Cr <sup>2+</sup> | $3d^4$        | 2   | 2   |
| Fe      | $3d^6 4s^2$    | Fe <sup>3+</sup> , Mn <sup>2+</sup> | $3d^5$        | 0   | 5/2 |
| Co      | $3d^7 4s^2$    | Co <sup>3+</sup> , Fe <sup>2+</sup> | $3d^6$        | 2   | 2   |
| Ni      | $3d^8 4s^2$    | Co <sup>2+</sup>                    | $3d^7$        | 3   | 3/2 |
| Cu      | $3d^{10} 4s^1$ | Ni <sup>2+</sup>                    | $3d^8$        | 3   | 1   |
| Zn      | $3d^{10} 4s^2$ | Cu <sup>2+</sup>                    | $3d^9$        | 2   | 1/2 |

# Summary

1. Spin-orbit interaction
2. Magnetism in quantum theory

## Chapter 2 Magnetism in localized systems

1. Spherical potential
2. Larmor precession
3. Magnetism of inert gas
4. LS multiplex ground state of open shell ions and Hund's rule